

Waarom moeilijk doen als het makkelijk(er) kan

Fast analysis of Gas Properties

ORBITAL GAS PT 2

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Recognise this ?



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Knowing (L)NG-properties

- Gas composition
 - C1 → Cx including CO₂ H₂ Air
- LHV/HHV
 - Lower /Higher heating value (with or without water condensing energy loss)
- Wobbe Index
 - $W(i)$ = relation between heating value and relative density of a gas



Why measuring wobble

- Custody transfer
 - Selling Heating value against a measured delivered amount
- Boiler control
 - Optimisation of burning gas (less or more O₂/fuel)
- Turbine control
 - Operation of turbine (knocking)

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Existing Techniques

- Oxygen Demand Analysers (Wobbe Index)
 - Pro Fast , Commonly known
 - Con Large footprint , Maintenance , Utility gasses

- Proces Gas Chromatograph
 - Pro Complete composition , accuracy
 - Con Speed (cyclic)
Carrier , calibration gasses needed
Outside installation (maintenance)



The Alternative

Orbital GAS PT 2

- Fast
- Accurate
- No utility gasses
- Small footprint
- Hardly no maintenance

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Prefer this ?



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GAS-PT Pipe Mounted



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Background to GasPT

- Developed by GL (formerly Advantica/British Gas R&D) in UK
- Licensed worldwide and exclusively to CUI Global inc./Orbital

- Inferential Gas Properties Transmitter

- GasPT1 measured speed of sound & thermal conductivity
- GasPT1 inferred 4-gas mixture of CH₄, C₃H₈, N₂ and CO₂
- Calculated RD, CV, Wobbe, Compressibility Factor to ISO6976

- GasPT2 has GasPT1 measurements with added CO₂ sensor unit
- GasPT2 infers a 5-gas mixture of CH₄, C₂H₆, C₃H₈, N₂ and CO₂
- More accurate calculation of gas properties
- Wider application across the complete range of natural gases

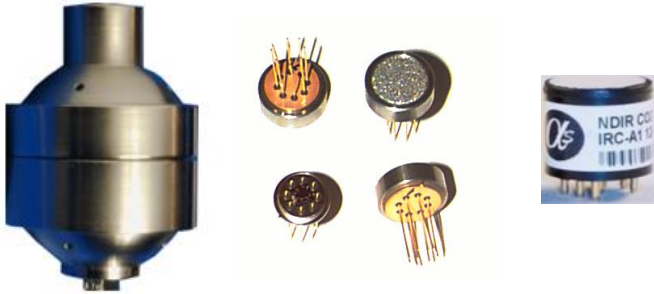
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Method of Operation

GasPT2 Directly Measures

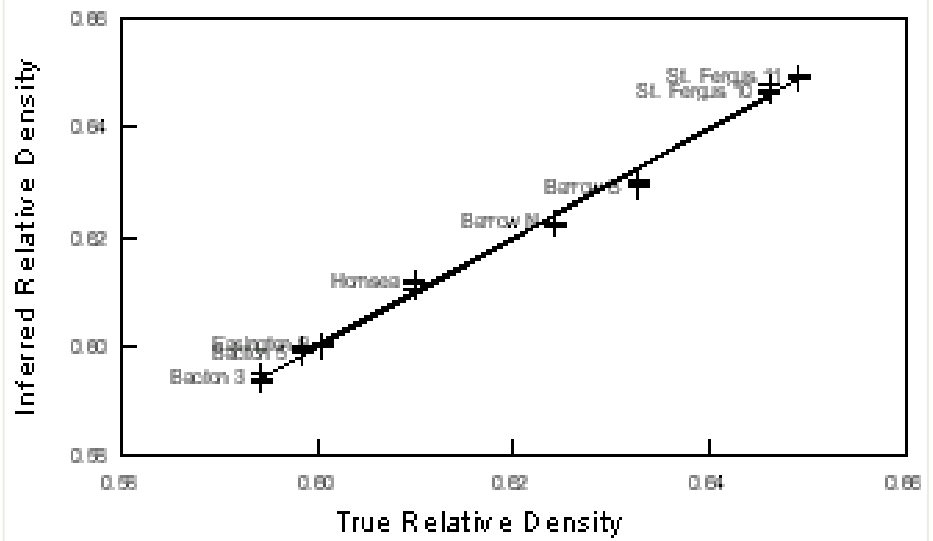
- Speed of Sound – Acoustic Resonator
 - Thermal Conductivity – TCD Sensor
 - Carbon Dioxide – NDIR Principle
- 
- Calibration gases are used to derive the response of each device
 - Algorithms solved and flashed to the GasPT2
 - 3 measurements are mathematically solved to derive a 4 component pseudo composition consisting of
 - Methane, Ethane, Propane, Nitrogen (Effective)
 - Plus Carbon Dioxide



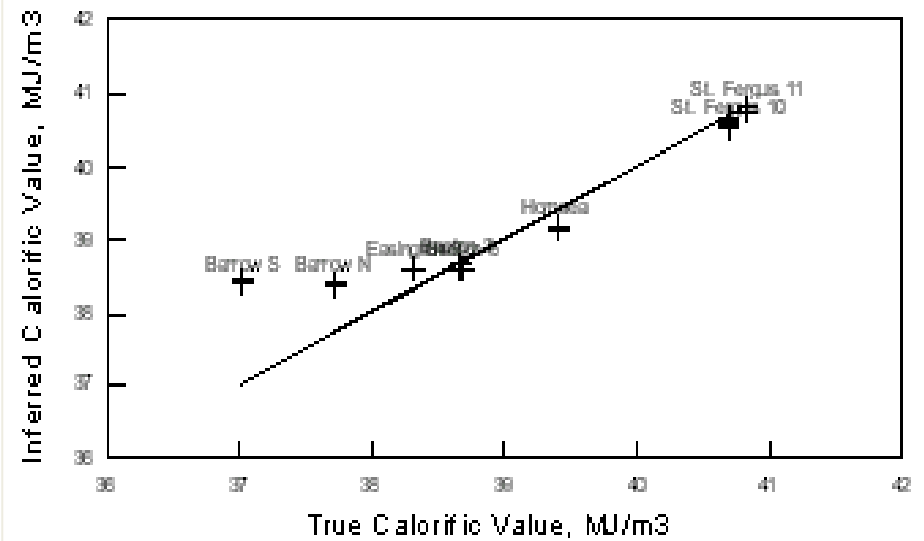
Principle of Operation

Speed of Sound Correlation

Relative Density Inference
(Prototype 1)



Calorific Value Inference
(Prototype 1)



Speed of Sound correlates strongly to molecular weight of gas

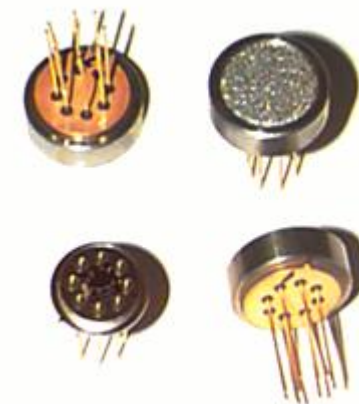
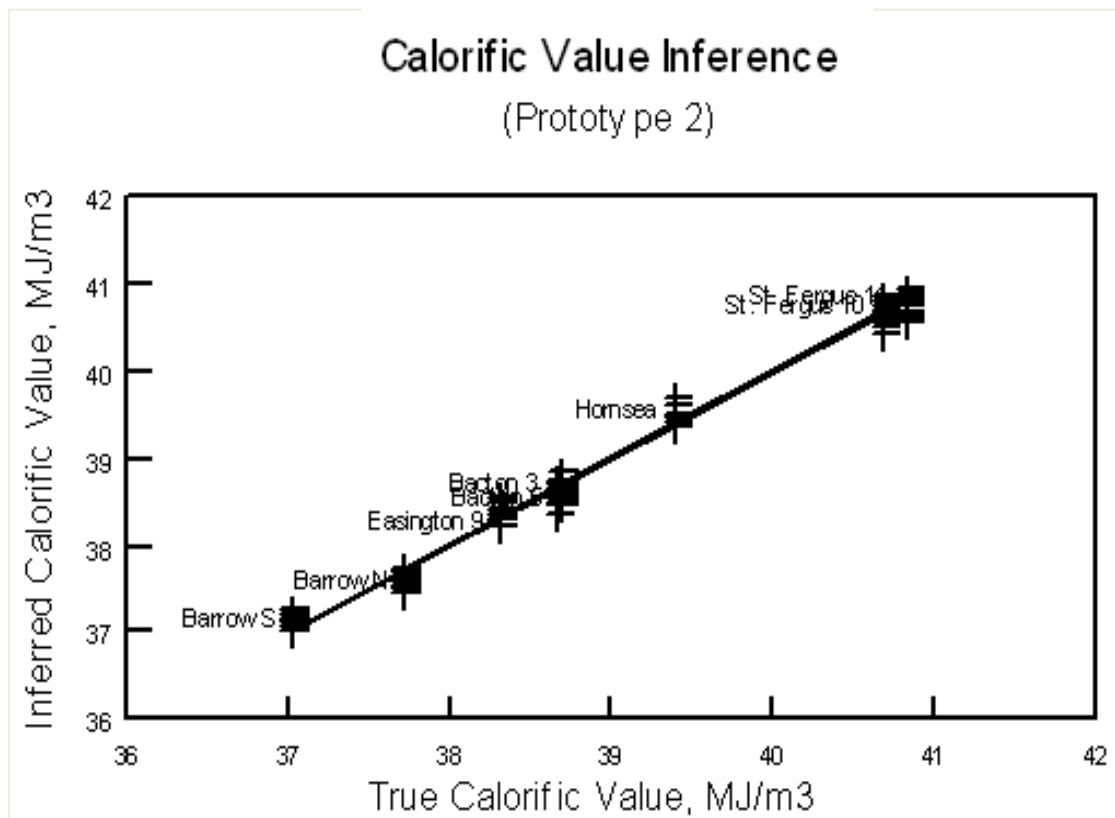
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Principle of Operation

Correlation using Thermal Conductivity



Thermal conductivity distinguishes between hydrocarbons and Inerts

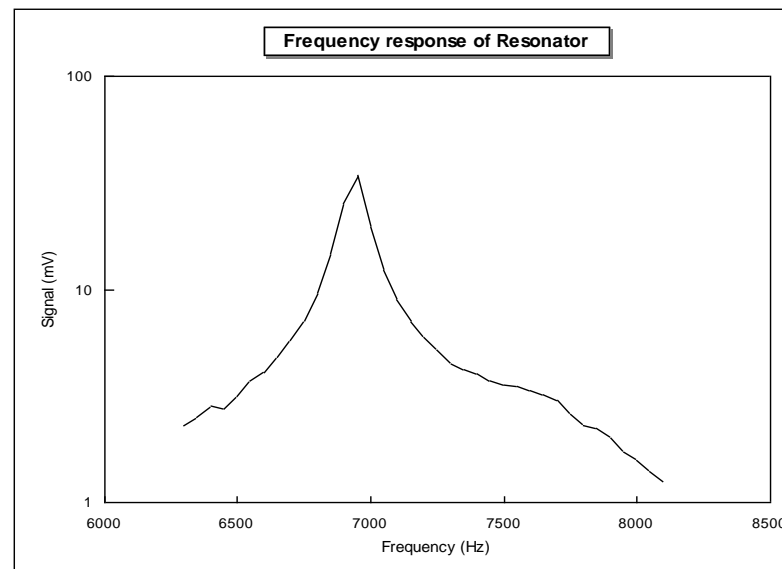
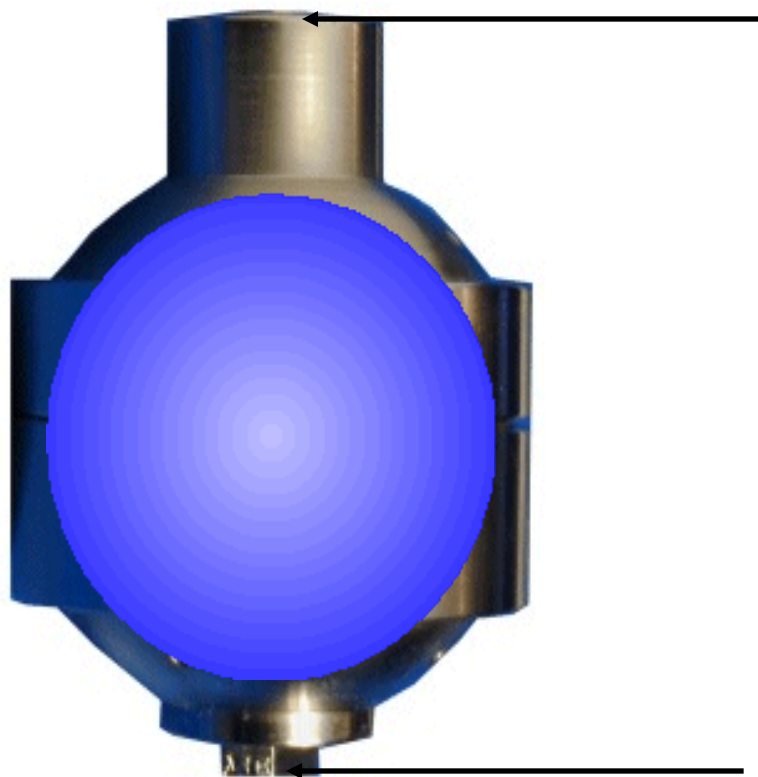
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GasPT: Speed of Sound

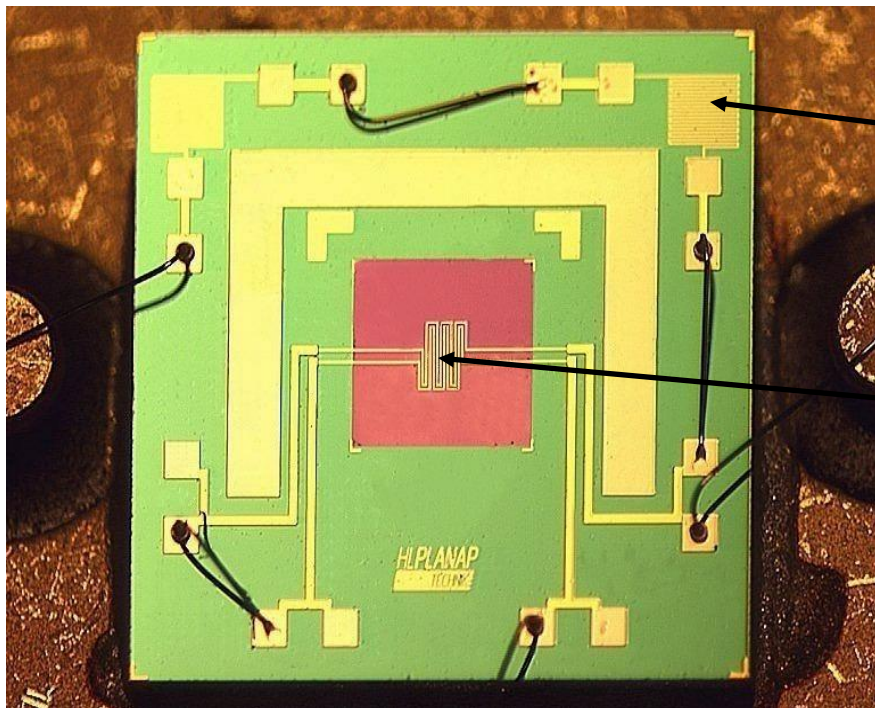
Microphone: Detects amplitude of oscillation



Miniature Speaker drives the resonance



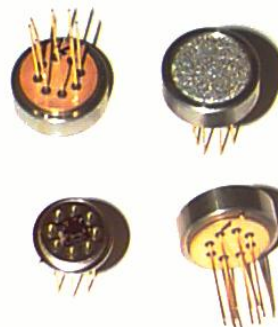
GasPT : Thermal conductivity (TCD)



Ambient temperature sensor

Thermal conductivity sensing resistor.

Heated above ambient



GasPT Sensor Units and Safety Interfaces

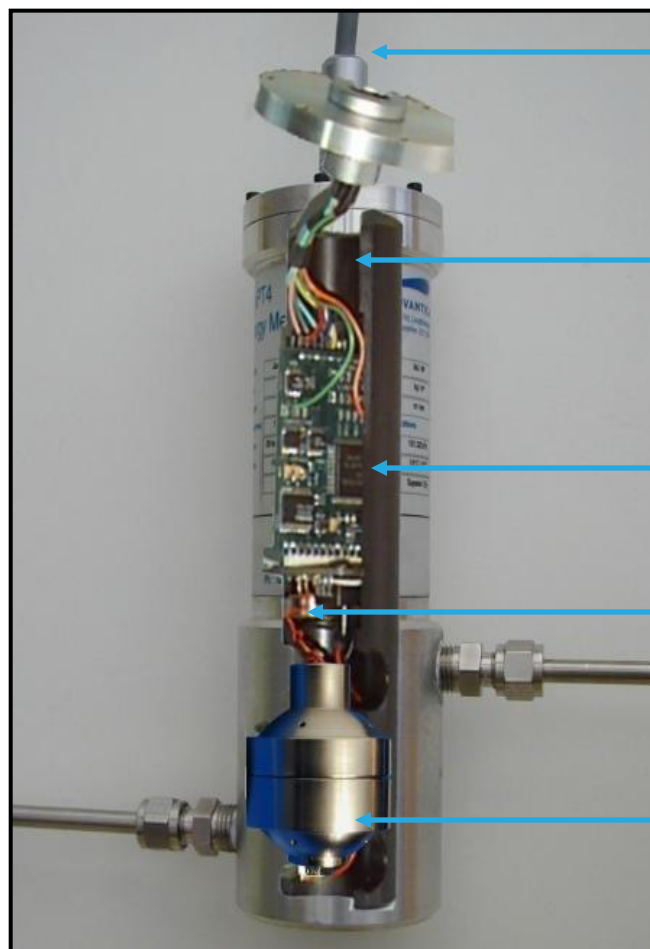


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Components



Electrical Connections

Insulating Sleeve

Processor Board

Thermal Conductivity
Sensor

Spherical Resonator



Theoretical Analysis

Gas Interchangeability

All hydrocarbons typically found in a natural gas can be represented by a mix of effective methane (CH₄'), effective ethane (C₂H₆') and effective propane (C₃H₈') that yield the same number of carbon and hydrogen atoms.

Examples :



(1 mol. Butane = 1 mol. Ethane + 1 mol. Propane - 1 mol. Methane)



(2 mols. Pentane = 2 mols. Ethane + 3 mols. Propane – 3 mols. Methane)



(1 mol. Hexane = 3 mols. Propane – 1 mol. Methane – 1 mol. Ethane)



Calculation of an inferred gas composition

Ref Gas	CH ₄	C ₂ H ₆	C ₃ H ₈	C ₄ H ₁₀	C ₅ H ₁₂	C ₆ H ₁₄	CO ₂	N ₂		Total
	94.451	3.11	0.512	0.198	0.17	0	0.647	0.912		100

Ref gas CV=38.731 MJ/m³

	Effective Methane			Effective Ethane			Effective Propane		
Methane	1.0	94.451	94.451	0.0	94.451	0	0.0	94.451	0
Ethane	0.0	3.11	0	1.0	3.11	3.11	0.0	3.11	0
Propane	0.0	0.512	0	0.0	0.512	0	1.0	0.512	0.512
Butane	-1.0	0.198	-0.198	1.0	0.198	0.198	1.0	0.198	0.198
Pentane	-1.5	0.17	-0.255	1.0	0.17	0.17	1.5	0.17	0.255
Hexane	-1.0	0	0	-1.0	0	0	3.0	0	0
			93.998			3.478			0.965

Pseudo Gas	CH ₄	C ₂ H ₆	C ₃ H ₈	C ₄ H ₁₀	C ₅ H ₁₂	C ₆ H ₁₄	CO ₂	N ₂		Total
	93.998	3.478	0.965	0	0	0	0.647	0.912		100

Inferred Gas CV=38.725 MJ/m³

Principle of Operation – Property Calculations

Physical Property Calculations	11 Components	5 Components	Δ Difference
Compressibility @ 15 °C	0.9978	0.9978	0.0000
Density @ 0 °C	0.7646	0.7646	0.0000
Relative Density @ 15 °C	0.5912	0.5912	0.0000
Gross CV @ 15 °C	38.7309	38.7334	-0.0025
Net CV @ 15 °C	34.9209	34.9234	-0.0025
Sooting Index	0.5130	0.5131	-0.0001
ICF	-0.1434	-0.1414	-0.0020
Wobbe @ 15 °C	50.3722	50.3755	-0.0033

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Operational Specification

Sensor Unit ambient or sample gas temperature	-20 to +55 DegC
Safety Interface ambient temperature	-20 to +50 DegC
Max. pressure	4.3 psig (~300mbar)
Humidity	Non-condensing
Sample gas flowrate	0.5 to 2.0 l/min
Analysis Time	Instrument Update < 2 sec
ATEX, IECEx and CSA approvals	Probes suitable for NG in Zone 1 and Zone 2 hazardous areas
Accuracy	Better than $\pm 0.3\%$ error on CV and Wobbe measured value
Repeatability	CV and Wobbe $\pm 0.04\text{MJ/m}^3$ (1.07Btu/ft ³) at Amb. Temperature
Power requirement	24 Vdc @ 250 mA (trimmed to 23Vdc)

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- Methane 50 - 100 %
- Ethane 0 - 12.0 %
- Propane 0 - 7.0 %
- iso-Butane 0 - 1.0 %
- n-Butane 0 - 1.0 %
- iso-Pentane 0 - 0.5 %
- n-Pentane 0 - 0.5%
- C6+ Total 0 - 0.5 %
- Carbon Dioxide 0 - 20 % (Adjustable zero)
- Nitrogen 0 - 20.0 %
- H₂S < 5ppm
- He < 0.5%
- O₂ and H₂ Low levels only
- H₂O Dry, non-condensing



GAS PT -2 outputs

- CV/BTU Value
- Relative Density
- Wobbe
- Compressibility
- Motor Octane Number
- Carbon Dioxide content



Conclusion

- A -2- second GAS PT measurement benefit in
 - Fast determination of gas properties
 - Easy installation requirement
 - Less maintenance
 - Certified for Custody Transfer
- Resulting in
 - Optimal use of burning or selling (L)NG in a easy way

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Vragen ?????



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